Phenylselenoazidation of Conjugated Enals. Allylic Rearrangement of 2-Alkylidene-3-azidopropanals and Propionic Acids

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 β -Azido α -phenylseleno aldehydes were isolated as adducts of PhSeN $_3$ on conjugated enals bearing an α -alkyl substituent. Their oxidation leads to 2-alkylidene-3-azido propanals and propionic acids. An allylic rearrangement of the azido group was observed.

In the preceding letter, $^{1)}$ we described the preparation of β -methoxy- β -hydroxy- and β -acetoxy- α -phenylseleno aldehydes by the reaction of conjugated enals involving the corresponding oxygen nucleophiles and an intermediate seleniranium cation resulting from the addition of PhSeCl across the C=C bond.

We were interested in analogous reactions using nitrogen nucleophiles. Previous studies deal with the phenylseleno-activated introduction of the nitrite anion on olefins, $^{2)}$ conjugated dienes $^{3)}$ and with that of the azide anion on simple olefins. $^{4-6)}$ It must also be noted that selenenamides add to Michael acceptors like conjugated ketones $^{7)}$ and enals. $^{8)}$

We present, here, our first results concerning the addition of $PhSeN_3$ on the enals $\underline{2}$ in dimethylsulfoxide 9 and some oxidative and reductive transformations of the adducts.

R
CHO
CHO
CHO
$$R^2$$
CHO
 R^1
 R^2
CHO
 R^2
CHO
 R^3
 R^4
CHO
 R^2
CHO
 R^2
CHO
 R^2
CHO
 R^3
 R^4
CHO
 R^2
 R^4
 R^4
 R^4
CHO
 R^2
 R^4
 $R^$

We first observed that enals $\underline{1}$ which bear no substituent on the α -carbon give complex mixtures probably containing elimination products of the two regioisomeric adducts. With enals $\underline{2}$, the addition products $\underline{3}$ were only formed as mixtures of diastereoisomers. From this result, we can suggest, as previously assumed in the case of oxygen nucleophiles, that the reactions involve the reversible formation of an intermediate seleniranium cation. The oily β -azido aldehydes $\underline{3}$ were readily prepared in very good yields (Table 1).

Table 1. $^{11}\beta$ -Azido-	x-phenylseleno	aldehydes	s <u>3</u> and	their	oxidation
	products 4, 5	, <u>6</u> , <u>7</u> , ar	nd <u>8</u>		

No. R ¹ (CH ₂ R')	_R 1	R ²	3 Yield Diget		Yield	(Isomer distrib.		Yield	Isomer distrib.	
	10	<u> </u>	Diast. comp.	11010 %	<u>4</u>	<u>5</u>	<u>6</u>	<u>%</u>	7	8	
a	Me	Н	91	-	58		50 50		85		
b	Me	Ме	90 -	62/38	60	10	90				
С	Me	Et	80	55/45	62	10	90		90	50	50
d	Me	Ph	93	58/42	85		80	20	75	67	33
е	Et	Ме	82	70/30	65	100			62		
f	-(CH ₂)	4	79	70/30	52	10	00		90		

The sodium periodate oxidation of the β -azido- α -phenylseleno aldehydes $\underline{3}$ was then achieved. 12) After syn-elimination reaction on the intermediate selenoxide, $\underline{3a}$ leads to a 1:1 mixture of the allylic azide $\underline{4a}$ ($\underline{5a}$) and the vinylic azide $\underline{6a}$. This result is identical to those concerning the oxidation of β -phenylseleno azides. The β -phenyl substituted aldehyde $\underline{3d}$ gives the vinylic azide $\underline{6d}$ and the rearranged allylic azide $\underline{5d}$ as major compound. The same oxidation of $\underline{3b}$ and $\underline{3c}$ gave mixtures of the first formed allylic compounds $\underline{4b}$ and $\underline{4c}$, respectively, and the rearranged ones $\underline{5b}$ and $\underline{5c}$ as major components. In the case of the selenides $\underline{3e}$ and $\underline{3f}$, the allylic azides of unique structure $\underline{4e}$ ($\underline{5e}$) and $\underline{4f}$ ($\underline{5f}$) were isolated with fair yields. (Table 1). The literature does not describe examples of azide allylic rearrangements on structure having a double bond conjugated with a carbonyl group.

As previously described for other structures, 1,14) the hydrogen peroxide oxidation of the α -phenylseleno aldehydes $\underline{3}$ allow the preparation of the α,β -unsaturated carboxylic acids $\underline{7}$ ($\underline{8}$). $\underline{7c}$ and $\underline{7d}$ rearrange partially into the allylic azides $\underline{8c}$ and $\underline{8d}$, respectively, as observed for the corresponding

aldehydes (Table 1). No traces of vinylic azides analogous to the enals $\underline{6}$ were found in the reaction mixtures.

To determine the stereochemistry of these unsaturated compounds $\underline{4}-\underline{8}$, a detailed NMR study is under investigation. At this time, we have not established if the rearranged allylic azides $\underline{5}$ correspond to a formal azidation of the α -alkyl group of the enals $\underline{2}$ with retention of the stereochemistry. With the aim to convert the azides $\underline{3}$ into primary amines and amides, we first protected the aldehyde function. The acetals $\underline{9}$ were prepared in good yields. The reduction step was achieved by the Staudinger method involving the hydrolysis of an intermediate iminophosphorane. The unstable amines $\underline{10}$, also obtained by another way, were treated with acetic anhydride in methanol. The acetamides $\underline{11}$ were isolated in correct yields.

$$R^{2}$$
 R^{1}
 R^{2}
 R^{2

The oxidative syn-elimination process achieved on the amide $11b^{20}$ gives the unsaturated acetal 12 in a good yield. Other transformations of structures 10 and 11 leading to α -aminomethyl α,β -unsaturated aldehyde and carboxylic acid derivatives, are now under investigation. The transformations of adducts resulting from the reaction of enals 2 with morpholino benzeneselenenamide is also being studied.

References

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- 9) PhSeCl (1.2 equiv.), NaN₃ (1.8 equiv.), DMSO, 20 °C, 16 h.
- 10) Acetoxyselenenylation of the same enals leads to α -seleno enals (Ref.1).
- 11) All the compounds isolated were purified by silicagel chromatography and characterized by $^1\mathrm{H}~$ and $^{13}\mathrm{C}~\mathrm{NMR}.$
- 12) NaIO₄, MeOH, H₂O, NaHCO₃, O °C \rightarrow 20 °C, 3 h. The vinylic azides <u>6a</u> and <u>6d</u> were separated chromatographically (silicagel, Petroleum ether/CH₂Cl₂ 80/20). The allylic azides <u>4a</u> (<u>5a</u>) and <u>4c</u> cannot be isolated in pure form.
- 13) Rearrangements of allylic azides: A. Gagneux, S. Winstein, and W. G. Young, J. Am. Chem.Soc., 82, 5956 (1960); C.A. Van Der Werf and V.L. Heasley, J. Org.Chem., 31, 3534 (1966); J. Cleophax, A. Olesker, A Rolland, S.D. Gero, and A. Forchioni, Tetrahedron, 33, 1303 (1977); I. Z. Kabore, Q. Kuong-Huu, and A. Pancrazi, ibid., 34, 2807 (1978); S.I. Murahashi, Y. Taniguchi, Y. Imada, and Y. Tanigawa, J. Org. Chem., 54, 3292 (1989).
- 14) ${\rm H_2O_2}$ 30%, ${\rm CH_2Cl_2/THF}$ (4/1), 10 °C \rightarrow + 20 °C, 16 h. See : F. Outurquin and C. Paulmier, Synthesis, 1989, 689 . The acids 7,8 were purified by basic treatment followed by a careful acidification. They are oily compounds except 7f (8f) : F = 121 °C.
- 15) MeOH, HC(OMe)₃, pTSA, molecular sieves, reflux 4 h. The acetals <u>9a</u>, <u>9c</u>, and <u>9d</u> are oils purified by chromatography and isolated as diastereomeric mixtures: <u>9a</u>: 69%; <u>9c</u>: 77%; <u>9d</u>: 64%.
- 16) PPh_3 , THF, reflux 3 h then H_2O . The amines $\underline{10}$ obtained as unstable oils are characterized by 1H NMR.
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- 18) SnCl₂, PhSH, THF, Et₃N, 20 °C, 0.5 h. See : M. Bartra, P. Romea, F.Urpi, and J. Vilarrasa, Tetrahedron, $\underline{46}$, 587 (1990).
- 19) Acetamides <u>11</u> are oily compounds separated from triphenylphosphine oxide by repeated extractions with warm hexane. Yields: <u>11b</u>: 62%; <u>11c</u>: 69%; <u>11d</u>: 71%.
- 20) Compound $\underline{12}$ is prepared (66% yield) by the meta-periodate oxidation of the acetamide $\underline{11}$ (R¹=R²=Me) (3 h, 20 °C. See Ref.1) and purified by silicagel the chromatography (elution : CH_2Cl_2).

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